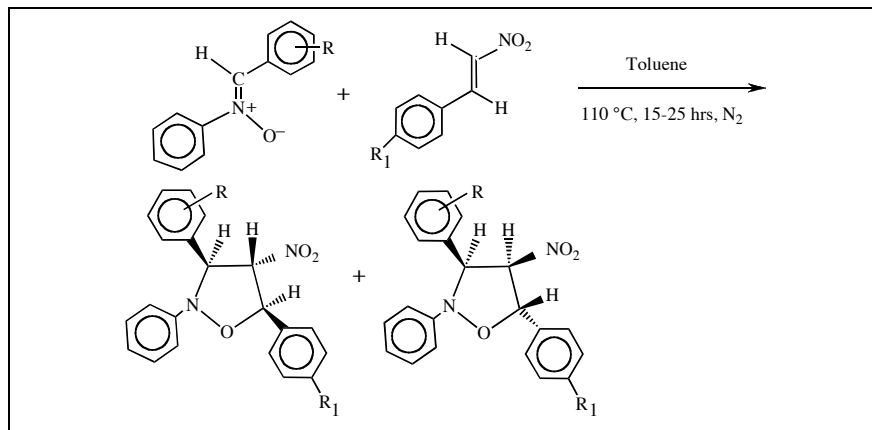


Avijit Banerji* [1], Maya Gupta [1], Pizush Kanti Biswas [1],
Thierry Prangé [2] and Alain Neuman [2]

[1] Centre of Advanced Studies on Natural Products including Organic Synthesis, Department of Chemistry,
University Colleges of Science and Technology, University of Calcutta, 92, Acharya Prafulla Chandra Road,
Kolkata-700009, India.

[2] Laboratoire de Chimie Structurale Biomoléculaire(UMR 7033CNRS), Université Paris Nord, 74 rue M.
Cachin F-93012 Bobigny, France.

Received October 17, 2006



Cycloadditions of *C,N*-diarylnitrones to β -nitrostyrenes occurred to yield two diastereoisomeric cycloadducts, the 3,4-*trans*-4,5-*trans* substituted isoxazolidine derivatives being formed selectively as the major products. These were characterised by spectroscopic and X-ray data. Conformational studies were carried out by X-ray crystallography and Molecular Modelling.

J. Heterocyclic Chem., **44**, 1045 (2007).

INTRODUCTION

Isoxazolidines can be regarded as templates for 1,3-difunctionalised compounds which can be used in the synthesis of several types of biologically active molecules [2-12]. These isoxazolidines become even more attractive as intermediates if a nitro group, which can be transformed into several functionalities, is present in the ring. With the objective of synthesising substituted 4-nitroisoxazolidines, we have carried out a series of 1,3-dipolar cycloadditions of nitrones with *trans*- β -nitrostyrene and its *p*-nitro derivative. Due to the presence of the conjugated nitro group, the β -nitrostyrenes have low-lying LUMOs, and these cycloadditions would be essentially of Sustmann Type I (Dipole_{HOMO}-Dipolarophile_{LUMO}) [13]. In view of the highly polarised nature of the dipolarophile frontier orbitals, a high degree of regioselectivity is expected to yield the desired 4-nitroisoxazolidines.

We report here the synthesis of a number of 3,4-*trans*-4,5-*trans*-3,5-diaryl-2-phenyl-4-nitroisoxazolidine cycloadducts with high regio- and stereo-selectivity. The structures and stereochemistry of these cycloadducts as well as the minor cycloadducts, with 3,4-*cis*-4,5-*trans* stereochemistry were settled by detailed spectroscopic,

and for selected products, - X-ray crystallographic, analysis. We were also interested in the conformational aspects of these compounds, and carried out computer-assisted energy minimisations.

RESULTS AND DISCUSSION

Cycloadditions of six *C*-aryl-*N*-phenyl aldonitrones (**1a-f**) with β -nitrostyrene **2a** and 4-nitro- β -nitrostyrene **2b** were investigated. Five of these nitrones (**1a-e**) had *C*-aryl rings with substituents varying from the electron withdrawing nitro-group to the electron-donating methoxy group, whereas the *C*-furyl-*N*-phenyl nitron was new.

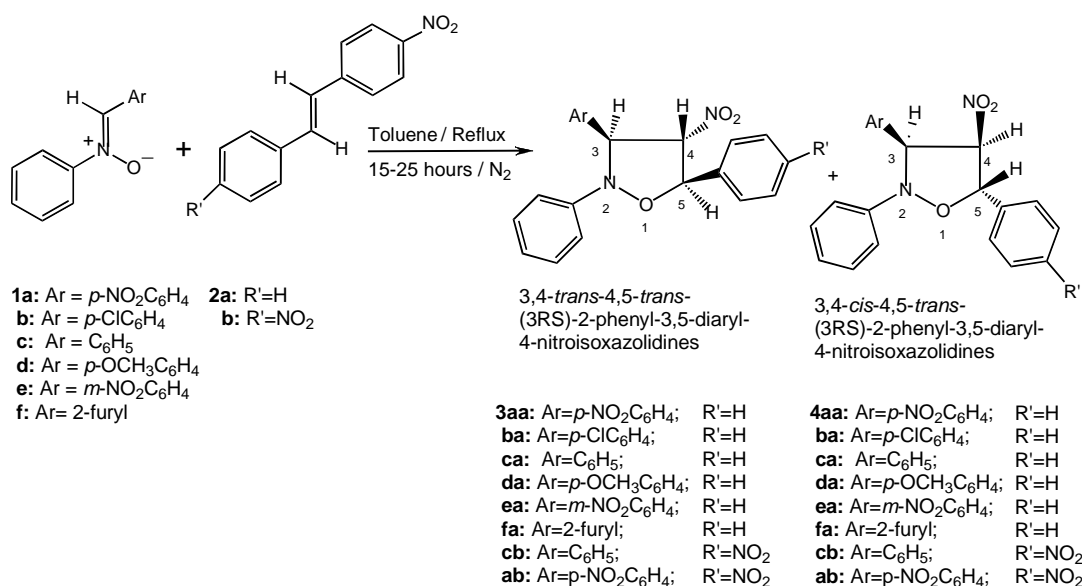
The cycloadditions were performed by refluxing the nitrones with equimolar amounts or with slight molar excess of the dipolarophiles under nitrogen atmosphere for 15-25 hrs. Conversions to products were usually above 80%. The 3,4-*trans*-4,5-*trans*-3,5-diaryl-2-phenyl-4-nitroisoxazolidine cycloadducts (Type A) were the major cycloadducts in all the reactions. The corresponding diastereoisomeric 3,4-*cis*-4,5-*trans*-3,5-diaryl-2-phenyl-4-nitroisoxazolidine cycloadducts (Type B) were obtained as the minor products in almost all the reactions. The product ratios of the cycloadducts were determined by ¹H NMR analysis of the crude reaction mixtures. The IR

spectra of all the isolated pure compounds in all series of cycloadducts showed the expected bands for the nitro group (*ca.* 1500-1550, 1350 cm^{-1}) and for the aromatic moieties. The UV spectra of the cycloadducts lacked the strong absorptions characteristic of the conjugated nitro groups, thus indicating that cycloadditions had occurred.

The EI-MS analysis of a typical example **3aa**, showed the molecular ion-peak at m/z 391 (87%). Other important peaks appeared at m/z 344 (8%; $\text{M}^+\text{-HNO}_2$), 285 (25%; $\text{M}^+\text{-PhCHO}$) and the fragments obtained from electron-impact induced cycloreversion at m/z 242 (10%), and 149 (7.5%). The loss of the elements of PhCHO, and a peak at m/z 238 from concomitant cleavage at N-2/C-3 and C-4/C-5, confirmed the presence of a phenyl group at C-5. The 300 MHz ^1H NMR spectra of the products belonging

of $J_{3,4}$ seemed to indicate a 3,4-*trans* orientation for **3aa**. The relative stereochemistry and conformation of **3aa** was confirmed by its X-ray crystallographic analysis [14]. The ORTEP projection of this molecule is shown in Figure 1. Compound **3aa** crystallised in the triclinic system with the space group P-1. The heterocyclic ring is in the distorted envelope conformation, with 3,4- and 4,5-substituents *trans*, as well as the nitrogen phenyl substituent with respect to the 3-aryl group. The oxygen forms the tip of the envelope. Cycloadducts **3** (**ba**, **ca**, **da**, **ea**, **fa**, **cb** and **ab**), which were the major cycloadducts in the reactions of **1(a-f)** respectively, showed ^1H and ^{13}C NMR characteristics very similar to those of **3aa**. Thus, all these belonged to the 3,4-*trans*-4,5-*trans*-series (Type A cycloadducts). Cycloadducts **3cb** and **3ab**, also belonging

Scheme-1



to both types A and B showed that H-3 and H-5 protons appeared as doublets and there was no coupling between them. Further, the H-3, H-5 protons showed long-range coupling with aromatic protons in COSY-LR spectrum, thereby showing their benzylic nature. The H-4 proton coupled to both H-3 and H-5 and appeared as double doublet. H-4 correlations to aromatic protons were not observed in the COSY-LR. This established the structure of the major isomers to be 2-phenyl-3,5-diaryl-4-nitrosoxazolidine derivatives. A typical representative of the major cycloadducts, *viz.* **3aa**, showed the presence of the H-3 doublet at δ 5.77 in its ^1H NMR spectrum. This was coupled to a double-doublet H-4 at δ 5.27, which in turn was coupled with the downfield H-5 doublet at δ 5.84; $J_{3,4}$ and $J_{4,5}$ were respectively 3.7 and 5.8 Hz. The magnitude

to Type A, were obtained from 4-nitro-*trans*- β -nitrostyrene **2b** and nitrones **1c** and **1a** respectively. The 300 MHz ^1H NMR spectrum of **3cb** in d_6 -DMSO showed that H-3 and H-5 protons appeared at δ 5.75 and δ 6.16 as doublets and the H-4 proton as δ 5.93 as a double doublet; $J_{3,4}$ and $J_{4,5}$ were respectively 3.2 and 4.2 Hz. The relative stereochemistry of **3cb** as 3,4-*trans*-4,5-*trans* was confirmed by X-ray crystallographic analyses [14]. This compound crystallised in the monoclinic form in the space group P2₁/n. The ORTEP diagram for **3cb** is shown in Figure 2. The five-membered heterocyclic ring exists as a twist envelope, in between the two idealised envelope conformations with the tips being at N2 and O1. Owing to significant ring puckering all substituents come in quasi-axial or quasi-equatorial positions. The nitrogen atom of

the isoxazolidine ring is pure pyramidal. In the relative configurations shown, the lone pair on N(2) is set above the plane by convention. A statistical disorder of the orientation of the nitro group at C(4) was observed in the crystal lattice at room temperature. This is a dynamic process but X-ray diffraction gives only an "average" static view of the phenomenon. During refinements, this group was modelled as a three static-component group (three positions with the sum of occupancies equal to one, for the two rotating oxygens, i.e. 0.4:0.35:0.25). Cycloadduct **3ab** showed similar spectroscopic properties and hence also belonged to Type A.

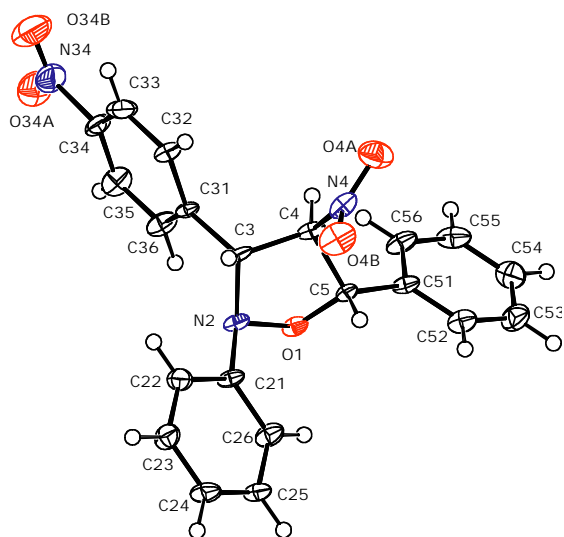


Figure 1. Structure of **3aa** as determined by X-Ray crystallographic analysis (ORTEP Projection).

The minor cycloadducts Type B were established to have 3,4-*cis*-4,5-*trans*-configuration. COSY-LR spectra of this type of cycloadducts showed long range correlations of H-3 and H-5 with aromatic protons. This established the overall structure as 3,4-*cis*-4,5-*trans*-3,5-diaryl-2-phenyl-4-nitroisoxazolidine. The signals of H-3, H-4, H-5 are situated close together. Cycloadducts **4(aa, ba, ca, da, ea, fa, cb and ab)** belong to this type of cycloadduct.

Conformational analysis and energy minimisation of the all-*trans* isomer along with its minor stereoisomer 3,4-*cis*, 3,4-*trans* have been carried out by taking advantage of the X-ray crystallographic data of the Type A isomers.

Energy minimisation of the 3,4-*trans*-4,5-*trans* isomer **3aa** (Type A), 3,4-*cis*-4,5-*trans* **4aa** (Type B) was carried out by conjugate gradient method using DISCOVER Molecular Simulation Program (Version 2.98) [15]. In all cases normal completion of conjugate gradient was achieved in less than 2000 steps of energy evaluations; the energy minimisation was stopped when the RMS derivative of energy reached a value of 0.001 kcal/mol.

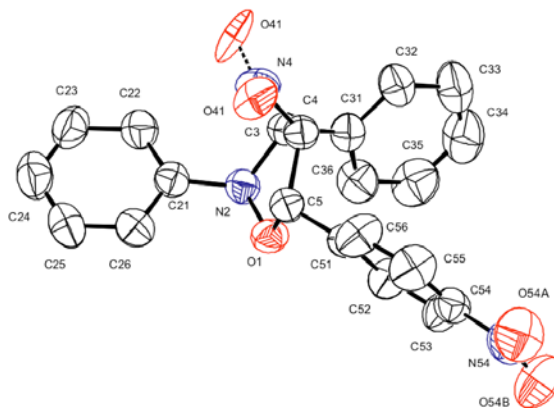


Figure 2. Structure of **3cb** as determined by X-Ray crystallographic analysis (ORTEP Projection). Only one rotamer of the nitro group at C4 is shown for clarity.

The relative potential energy of **4aa** compared to **3aa** was calculated to be 2.15 kcal/mol. The X-ray structure of all-*trans* isomer (compound **3aa**) has close similarity with the model structure of minimum energy.

The reaction of the new nitrono *C*-furfuryl-*N*-phenyl nitrono (**1f**) with *trans* β -nitrostyrene was carried out by refluxing for 16 hrs. in dry toluene under nitrogen atmosphere. The major product was confirmed as **3fa**, i.e. 3,4-*trans*-4,5-*trans* isomer from its NMR characteristics. Thus the selectivity of the cycloaddition process of this *C*-2-furfuryl nitrono was similar to the *C*-phenyl nitrono. This nitrono was prepared by reacting furfural with *N*-phenylhydroxylamine.

EXPERIMENTAL

General. Melting points were recorded on an electrically heated Kofler Block and are uncorrected. UV spectra were recorded on a Hitachi U-3501 spectrophotometer in 95% aldehyde free spectroscopic ethanol. IR spectra were recorded on KBr disc or as a thin film on a Perkin-Elmer FT-IR model. Mass spectra were recorded on a JEOL JMS-600 Mass spectrometer. Column chromatography (CC) was carried out using neutral alumina and silica gel. TLC was performed using silica gel G. Petrol refers to the fraction, b.p. 60-80°. Anhydrous sodium sulphate was used for drying extracts. Analytical samples were routinely dried over anhydrous calcium chloride *in vacuo* at room temperature. The starting materials prepared were identified by comparison of their melting points with those reported in the literature; as well as from elemental analyses and from IR spectra. NMR spectra were recorded in CDCl₃ or in d₆-DMSO solution on a Bruker AM-300L equipped with an ASPECT-3000 computer and an Array Processor using a switchable dual ¹H-¹³C 5 mm dual probe (normal and inverse) at ambient temperature. X-Ray crystallographic data were recorded at room temperature at the LURE synchrotron facility, Orsay, France (MoK α λ = 0.953 Å) for **3aa**, and on a PHILIPS PW 1100 diffractometer (operating the CuK α radiation, λ = 1.5418 Å) for **3cb**. The two structures were solved using the SHELXS program [16] and anisotropically refined using SHELXL [17].

General procedure for the Cycloaddition Reactions. A solution of the nitron (3.5-5.4 mM) in anhydrous toluene (10-15 ml) was added at a time to a solution of the dipolarophile (usually 1 to 1.5 molar proportions) in anhydrous toluene (10-15 ml). The reaction mixture was refluxed for several hours under nitrogen atmosphere. The progress of the reaction was monitored by TLC and 300 MHz ^1H NMR. After the reaction period solvent was removed from the reaction mixture by distillation under reduced pressure in a Büchi rotary evaporator. The residue was analysed by 300 MHz ^1H NMR and then subjected to Column Chromatography over silica gel, and sometimes over neutral alumina.

Reaction of C-(4-nitrophenyl)-N-phenyl nitron with β -nitrostyrene).

3RS(3R*,4S*,5R*)-2,5-Diphenyl-3-(4-nitro-phenyl)-4-nitro isoxazolidine (3aa). Light yellowish crystals, m.p. 126 °C (yield 84%) from 30% benzene in petrol; IR (KBr): 1556, 1347, 834, 750, 693 cm^{-1} ; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.26 (1H, dd, $J=3.8, 5.9, \text{H-4}$), 5.77 (1H, d, $J=3.8, \text{H-3}$), 5.84 (1H, d, $J=5.9, \text{H-5}$), ~7.02 (1H, obscured by H-2, H-6, $\text{H}_{\text{A-4}}$), 7.09 (2H, d, $J=7.9, \text{H}_{\text{A-2,6}}$), 7.34 (2H, dis. dd, $\text{H}_{\text{A-3,5}}$), 7.37-7.42 (5H, br. s, $\text{H}_{\text{C-2,6,3,5,4}}$), 7.77 (2H, d, $J=8.8, \text{H}_{\text{B-2,6}}$), 8.28 (2H, d, $J=8.8, \text{H}_{\text{B-3,5}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 73.5 (C-3), 83.6 (C-5), 101.4 (C-4), 115.3 ($\text{C}_{\text{A-2,6}}$), 124.0 ($\text{C}_{\text{A-4}}$), 124.9 ($\text{C}_{\text{B-3,5}}$), 127.0 ($\text{C}_{\text{C-2,6}}$), 128.2 ($\text{C}_{\text{B-2,6}}$), 129.6 ($\text{C}_{\text{C-3,5}}$), 129.9 ($\text{C}_{\text{A-3,5}}$), 130.0 ($\text{C}_{\text{C-4}}$), 135.0 ($\text{C}_{\text{C-1}}$), 145.7 ($\text{C}_{\text{B-1}}$), 148.5 ($\text{C}_{\text{A-1}}$), 148.8 ($\text{C}_{\text{B-4}}$), EI-MS: m/z 391 (M^+), 285, 238, 226, 192, 165, 105, and 77. Anal. Calcd for $\text{C}_{21}\text{H}_{17}\text{N}_3\text{O}_5$: C, 64.44; H, 4.34; N, 10.73 %. Found: C, 64.29; H, 4.43; N, 10.92%.

X-ray data. Triclinic, P-1 ($Z=2$) with $a=7.242(2)$; $b=8.371(2)$; $c=16.587(2)$ Å; $\alpha=84.32(4)$; $\beta=78.72(4)$ and $\gamma=70.01(4)^\circ$. Final R (1639 observed F) = 0.079 and R (all 1652 F data) = 0.085. Energy minimisations done by DISCOVER Molecular Simulation Program (Version 2.98) [15] for **3aa** and **4aa** were accurate to the extent of ± 0.2 kcal/mole.

3RS(3R*,4R*,5S*)-2,5-Diphenyl-3-(4-nitrophenyl)-4-nitro isoxazolidine (4aa). Canary yellow crystals, m.p. 192 °C (yield 16%) from 75% benzene in petrol. IR (KBr): 1552, 1347, 837, 749, 695 cm^{-1} ; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.41 (2H, m, H-3,4), 5.75 (1H, d, $J=3.3, \text{H-5}$), 7.09 (2H, d, $J=8.0, \text{H}_{\text{A-2,6}}$), 7.13 (1H, t, 7.4, $\text{H}_{\text{A-4}}$), 7.30 (2H, br.t, $J \sim 7.8, \text{H}_{\text{A-3,5}}$), 7.40 (5H, br. s, $\text{H}_{\text{C-2,6,3,5,4}}$), 7.76 (2H, d, $J=8.6, \text{H}_{\text{B-2,6}}$), 8.32 (2H, d, $J=8.6, \text{H}_{\text{B-3,5}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 73.7 (C-3), 81.8 (C-5), 99.6 (C-4), 118.4 ($\text{C}_{\text{A-2,6}}$), 125.1 ($\text{C}_{\text{B-3,5}}$), 125.4 ($\text{C}_{\text{A-4}}$), 126.9 ($\text{C}_{\text{C-2,6}}$), 128.8 ($\text{C}_{\text{B-2,6}}$), 129.2 ($\text{C}_{\text{A-3,5}}$), 129.4 ($\text{C}_{\text{C-3,5}}$), 130.2 ($\text{C}_{\text{C-1}}$), 130.2 ($\text{C}_{\text{C-4}}$), 148.2 ($\text{C}_{\text{A-1}}$), 148.2 ($\text{C}_{\text{B-4}}$), not found ($\text{C}_{\text{B-1}}$). Anal. Calcd. for $\text{C}_{21}\text{H}_{17}\text{N}_3\text{O}_5$: C, 64.44; H, 4.34; N, 10.73%. Found: C, 64.46; H, 4.48; N, 10.80%.

Reaction of C-(4-chlorophenyl)-N-phenyl nitron with β -nitrostyrene.

3RS(3R*,4S*,5R*)-2,5-Diphenyl-3-(4-chlorophenyl)-4-nitro isoxazolidine (3ba). Yellow microcrystalline solid (yield 82%) was obtained from 20% benzene in petrol. IR (KBr): 1555, 1361, 834, 785, 695 cm^{-1} ; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.26 (1H, dd, $J=4.0, 5.7, \text{H-4}$), 5.57 (1H, d, $J=4.0, \text{H-3}$), 5.78 (1H, d, $J=5.7, \text{H-5}$), 7.08 (2H, d, $J=7.9, \text{H}_{\text{A-2,6}}$), 7.29-7.49 (3H, m, $\text{H}_{\text{A-3,4,5}}$), 7.35 (2H, d, $J=8.2, \text{H}_{\text{B-2,6}}$), 7.37-7.42 (br. s, $\text{H}_{\text{C-2,3,4,5,6}}$), 7.47 (2H, d, $J=8.2, \text{H}_{\text{B-3,5}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 73.4 (C-3), 83.1 (C-5), 101.5 (C-4), 115.3 ($\text{C}_{\text{A-2,6}}$), 123.4 ($\text{C}_{\text{A-4}}$), 126.6 ($\text{C}_{\text{C-2,6}}$), 128.3 ($\text{C}_{\text{B-2,6}}$), 129.0 ($\text{C}_{\text{B-3,5}}$), 129.3 ($\text{C}_{\text{C-3,5}}$), 129.3

($\text{C}_{\text{C-4}}$), 129.5 ($\text{C}_{\text{A-3,5}}$), 134.6 ($\text{C}_{\text{C-1}}$), 135.3 ($\text{C}_{\text{B-4}}$), 136.7 ($\text{C}_{\text{B-1}}$), 148.9 ($\text{C}_{\text{A-1}}$). Anal. Calcd. for $\text{C}_{21}\text{H}_{17}\text{N}_2\text{O}_3\text{Cl}$: C, 66.23; H, 4.50; N, 7.36%. Found: C, 66.10; H, 4.66; N, 7.45%.

Reaction of C-phenyl-N-phenyl nitron with β -nitrostyrene.

3RS(3R*,4S*,5R*)-2,3,5-Triphenyl-4-nitroisoxazolidine (3ca). Yellow amorphous solid (yield 86%) was obtained from 30% benzene in petrol. IR (KBr): 1554, 1360, 756, 697; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.32 (1H, dd, $J=4.2, 5.9, \text{H-4}$), 5.60 (1H, d, $J=4.2, \text{H-3}$), 5.78 (1H, d, $J=5.9, \text{H-5}$), 7.04 (H, t, $J=7.5$), 7.10 (2H, d, $J=8.4, \text{H}_{\text{A-2,6}}$), 7.30 (2H, t, $J=8.0, \text{H}_{\text{A-3,5}}$), 7.36-7.44 (other Aromatic-H), 7.53 (2H, d, $J=6.5, \text{H}_{\text{B-2,6}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 74.2 (C-3), 83.2 (C-5), 101.7 (C-4), 115.3 ($\text{C}_{\text{A-2,6}}$), 129.3 ($\text{C}_{\text{A-3,5}}$), 123.1 ($\text{C}_{\text{A-4}}$), 126.6 ($\text{C}_{\text{C-2,6}}$), 126.7 ($\text{C}_{\text{B-2,6}}$), 128.6 ($\text{C}_{\text{B-4}}$), 129.0 ($\text{C}_{\text{C-3,5}}$), 129.2 ($\text{C}_{\text{B-3,5}}$), 129.2 ($\text{C}_{\text{C-4}}$), 135.5 ($\text{C}_{\text{C-1}}$), 138.2 ($\text{C}_{\text{B-1}}$), 149.2 ($\text{C}_{\text{A-1}}$). Anal. Calcd. for $\text{C}_{21}\text{H}_{18}\text{N}_2\text{O}_3$: C, 72.81; H, 5.24; N, 8.09%. Found: C, 72.93; H, 5.10; N, 8.01%.

Reaction of C-(4-methoxyphenyl)-N-phenyl nitron with β -nitrostyrene

3RS(3R*,4S*,5R*)-2,5-Diphenyl-3-(4-methoxyphenyl)-4-nitro isoxazolidine (3da). Orange red amorphous solid (yield 72%) was obtained from 30% benzene in petrol; IR (KBr): 1554, 1362, 836, 759, 696 cm^{-1} ; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.30 (1H, dd, $J=4.4, 5.8, \text{H-4}$), 5.49 (1H, d, $J=4.4, \text{H-3}$), 5.77 (1H, d, $J=5.8, \text{H-5}$), 6.90 (2H, d, $J=8.7, \text{H}_{\text{B-3,5}}$), 7.06 (1H, dis. t, $\text{H}_{\text{A-4}}$), 7.09 (2H, d, $J=8.0, \text{H}_{\text{A-2,6}}$), 7.30 (2H, d, $J=8.0, \text{H}_{\text{A-3,5}}$), obscured (2H, $\text{H}_{\text{B-2,6}}$), 7.38-7.45 (5H, br. s, $\text{H}_{\text{C-2,6,3,5,4}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 74.1 (C-3), 83.0 (C-5), 101.8 (C-4), 114.8 ($\text{C}_{\text{B-3,5}}$), 115.6 ($\text{C}_{\text{A-2,6}}$), 123.2 ($\text{C}_{\text{A-4}}$), 126.6 ($\text{C}_{\text{C-2,6}}$), 128.1 ($\text{C}_{\text{B-2,6}}$), 129.0 ($\text{C}_{\text{C-4}}$), 129.1 ($\text{C}_{\text{A-3,5}}$), 129.1 ($\text{C}_{\text{C-3,5}}$), 130.0 ($\text{C}_{\text{B-1}}$), 135.9 ($\text{C}_{\text{C-1}}$), 149.3 ($\text{C}_{\text{A-1}}$), 159.8 ($\text{C}_{\text{B-4}}$). Anal. Calcd. for $\text{C}_{22}\text{H}_{20}\text{N}_2\text{O}_4$: C, 70.20; H, 5.36; N, 7.44%. Found: C, 70.08; H, 5.43; N, 7.58%.

Reaction of C-(3-Nitrophenyl)-N-phenyl nitron with β -nitrostyrene.

3RS(3R*,4S*,5R*)-2,5-Diphenyl-3-(3-nitrophenyl)-4-nitro isoxazolidine (3ea). Yellow amorphous solid (yield 68%) was obtained from 30% benzene in petrol; IR (KBr): 1541, 1352, 812, 756, 693 cm^{-1} ; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.28 (1H, dd, $J=3.65, 5.9, \text{H-4}$), 5.76 (1H, d, $J=3.65, \text{H-3}$), 5.83 (1H, d, $J=5.9, \text{H-5}$), 7.07 (1H, t, 7.4, $\text{H}_{\text{A-4}}$), 7.10 (2H, d, $J=7.8, \text{H}_{\text{A-2,6}}$), 7.33 (2H, dd, 7.8, 7.4 $\text{H}_{\text{A-3,5}}$), 7.38-7.40 (5H, brd. s, $\text{H}_{\text{C-2,6,3,5,4}}$), 7.60 (1H, t, ~8.0, $\text{H}_{\text{B-5}}$), 7.89 (1H, d, 7.7, $\text{H}_{\text{B-6}}$), 8.21 (1H, dd, $J=1.1, 8.2, \text{H}_{\text{B-4}}$), 8.45 (1H, s, $\text{H}_{\text{B-2}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 73.4 (C-3), 83.7 (C-5), 101.4 (C-4), 115.4 ($\text{C}_{\text{A-2,6}}$), 122.4 ($\text{C}_{\text{B-2}}$), 124.0 ($\text{C}_{\text{A-4}}$), 124.1 ($\text{C}_{\text{B-4}}$), 127.0 ($\text{C}_{\text{C-2,6}}$), 129.6 ($\text{C}_{\text{C-3,5}}$), 129.9 ($\text{C}_{\text{A-3,5}}$), 130.0 ($\text{C}_{\text{C-4}}$), 130.8 ($\text{C}_{\text{B-5}}$), 133.2 ($\text{C}_{\text{B-6}}$), 135.0 ($\text{C}_{\text{C-1}}$), 141.0 ($\text{C}_{\text{B-1}}$), 148.9 ($\text{C}_{\text{A-1}}$), 149.3 ($\text{C}_{\text{B-3}}$). Anal. Calcd. for $\text{C}_{21}\text{H}_{17}\text{N}_3\text{O}_5$: C, 64.44; H, 4.34; N, 10.73%. Found: C, 64.55; H, 4.45; N, 10.64%.

3RS(3R*,4R*,5S*)-2,5-Diphenyl-3-(3-nitrophenyl)-4-nitro isoxazolidine (4ea). Yellow amorphous solid (yield 18%) from 75% benzene in petrol. IR (KBr): 1538, 1347, 812, 748, 693 cm^{-1} ; ^1H NMR (CDCl_3 , δ , 300 MHz): 5.42 (2H, m, H-3,4), 5.78 (1H, d, 5.5, H-5), 7.11 (2H, d, $J=7.0, \text{H}_{\text{A-2,6}}$), 7.14 (1H, t, 7.0, $\text{H}_{\text{A-4}}$), 7.30 (2H, t, $J=7.5, \text{H}_{\text{A-3,5}}$), 7.39-7.44 (5H, brd. s, $\text{H}_{\text{C-2,6,3,5,4}}$), 7.65 (H, t, ~8.0, $\text{H}_{\text{B-5}}$), 7.87 (H, d, 7.5, $\text{H}_{\text{B-6}}$), 8.28 (H, dd, $J=1.5, 7.5, \text{H}_{\text{B-4}}$), 8.47 (H, br. t, 1.5 $\text{H}_{\text{B-2}}$); ^{13}C NMR (CDCl_3 , δ , 75.5 MHz): 73.7 (C-3), 81.8 (C-5), 99.7 (C-4), 118.6 ($\text{C}_{\text{A-2,6}}$), 122.8 ($\text{C}_{\text{B-2}}$), 124.6 ($\text{C}_{\text{B-4}}$), 125.4 ($\text{C}_{\text{A-4}}$), 126.9 ($\text{C}_{\text{C-2,6}}$), 129.2

(C_{A-3,5}), 129.5 (C_{C-3,5}), 130.1 (C_{C-4}), 131.1 (C_{B-5}), 131.4 (C_{C-1}), 133.8 (C_{B-6}), 140.1 (C_{B-1}), 148.3 (C_{A-1}), 149.5 (C_{B-3}). *Anal.* Calcd. for C₂₁H₁₇N₃O₅: C, 64.44; H, 4.34; N, 10.73%. Found: C, 64.46; H, 4.48; N, 10.80%.

Ratio of **3ea**:**4ea** in crude reaction mixture by ¹H NMR - 77:23

Reaction of C-(2-Furyl)-N-phenyl nitrone with β-nitrostyrene.

3RS(3R*,4S*,5R*)-2,5-Diphenyl-3-furyl-4-nitroisoxazolidine (3fa). Yellow amorphous solid (yield 68.5%) was obtained from 10% benzene in petrol. IR (KBr): 1556, 1361, 754, 695 cm⁻¹; ¹H NMR (CDCl₃, δ, 300 MHz): 5.54 (1H, dd, J=3.3, 6.0, H-4), 5.71 (1H, d, J=6.0, H-5), 5.77 (1H, d, J=3.3, H-3), 6.39 (1H, dd, J=3.1, 1.8, H_{B-4}), 6.45 (1H, d, 3.1, H_{B-3}), 7.07 (1H, t, 7.4, H_{A-4}), 7.15 (2H, d, J=7.8, H_{A-2,6}), 7.34 (2H, dd, 7.8, 7.4, H_{A-3,5}), 7.40-7.45 (3H, m, H_{C-3,5,4}), 7.46 (1H, dis. d, H_{B-5}), 7.50 (2H, dd, J=7.6, 1.8, H_{B-2,6}): ¹³C NMR (CDCl₃, δ, 75.5 MHz): 68.6 (C-3), 84.0 (C-5), 98.1 (C-4), 109.7 (C_{B-3}), 111.2 (C_{B-4}), 115.7 (C_{A-2,6}), 123.8 (C_{A-4}), 127.4 (C_{C-2,6}), 129.4 (C_{A-3,5}), 129.8 (C_{C-4}), 129.6 (C_{C-3,5}), 135.5 (C_{C-1}), 143.8 (C_{B-5}), 148.8 (C_{A-1}), 150.0 (C_{B-2}). *Anal.* Calcd. for C₁₉H₁₆N₂O₄: C, 67.84; H, 4.79; N, 8.33%. Found: C, 67.65; H, 4.66; N, 8.00%.

3RS(3R*,4R*,5S*)-2,5-Diphenyl-3-furyl-4-nitroisoxazolidine (4fa). Canary yellowish amorphous solid (yield 25%) was obtained from 30% benzene in petrol. IR (KBr): 1555, 1370, 745, 695 cm⁻¹; ¹H NMR (CDCl₃, δ, 300 MHz): 5.15 (1H, d, 8.9, H-5), 5.44 (1H, dd, J=7.1, 8.9, H-4), 6.09 (1H, d, J=7.1, H-3), 6.37 (1H, dd, J=3.2, 1.9, H_{B-4}), 6.51 (1H, d, 3.3, H_{B-3}), 7.07 (1H, dd, 8.2, 1.0, H_{A-4}), 7.09 (2H, d, J=7.4, H_{A-2,6}), 7.28 (2H, t, J=7.5, H_{A-3,5}), 7.40-7.51 (5H, m, H_{C-2,6,3,5,4}), 7.48 (H, d, 3.3, H_{B-5}). ¹³C NMR (CDCl₃, δ, 75.5 MHz): 67.6 (C-3), 81.2 (C-5), 95.1 (C-4), 111.5 (C_{B-3}), 111.9 (C_{B-4}), 118.5 (C_{A-2,6}), 125.0 (C_{A-3,5}), not found (C_{A-4}), not found (C_{B-2}), not found (C_{C-1}), 127.1 (C_{C-2,6}), 129.5 (C_{C-3,5}), 129.8 (C_{C-4}), 144.2 (C_{B-5}), 149.2 (C_{A-1}). *Anal.* Calcd. for C₁₉H₁₆N₂O₄: C, 67.84; H, 4.79; N, 8.33%. Found: C, 67.65; H, 4.76; N, 8.20%.

Reaction of C,N-Diphenyl nitrone with 4-nitro β-nitrostyrene.

3RS(3R*,4S*,5R*)-2,3-Diphenyl-5-(4-nitrophenyl)-4-nitroisoxazolidine (3cb). Yellow crystals, m.p.170° (yield 40%) was obtained from 50% benzene in petrol. IR (KBr): 1550, 1344, 849, 754, 693 cm⁻¹; ¹H NMR (CDCl₃, δ, 300 MHz): 5.75 (1H, d, J=3.2, H-3), 5.93 (1H, dd, J=3.2, 4.2, H-4), 6.16 (1H, d, J=4.2, H-5), 7.01 (2H, t, 7.3, H_{A-3,5}), 7.11 (2H, d, J=8.5, H_{B-3,5}), 7.27 (2H, d, J=8.0, H_{B-2,6}), 7.29-7.32 (1H, m, H_{B-4}), 7.37 (1H, d, J=7.6, H_{A-4}), 7.45 (2H, d, J=7.7, H_{A-2,6}), 7.75 (2H, d, J=8.7, H_{C-2,6}), 8.24 (2H, d, J=8.7, H_{B-3,5}); ¹³C-NMR (d₆-DMSO, δ, 75.5 MHz): 72.7 (C-3), 80.8 (C-5), 99.6 (C-4), 116.1 (C_{A-2,6}), 123.4 (C_{A-4}), 123.8 (C_{C-3,5}), 127.2 (C_{B-3,5}), 128.2 (C_{C-2,6}), 128.4 (C_{B-4}), 129.0 (C_{B-2,6}), 129.2 (C_{A-3,5}), 137.5 (C_{B-1}), 143.8 (C_{C-1}), 147.7 (C_{A-1}), 148.3 (C_{C-4}). *Anal.* Calcd. for C₂₁H₁₇N₃O₅: C, 64.44; H, 4.34; N, 10.73%. Found: C, 64.24; H, 4.43; N, 10.67%.

X-ray data. Monoclinic, space group P2₁/n (Z=4) with a=11.140(3); b=18.123(3); c=10.113(3) Å and β=110.99(5)°.

Final *R* (2743 observed *F*) = 0.064 and *R* (all 3149 *F* data) = 0.081

Reaction of C-(4-Nitrophenyl)-N-(4'-chlorophenyl) nitrone with 4-nitro β-nitrostyrene.

3RS(3R*,4S*,5R*)-2-(Chlorophenyl)-3-(4-nitrophenyl)-4-nitro-5-phenyl isoxazolidine (3ab). Canary yellow crystals, m.p. 212° (yield 87%) was obtained from 50% benzene in petrol. IR (KBr) 1519, 1346, 841, 687 cm⁻¹; ¹H NMR (CDCl₃, δ, 300 MHz): 6.07 (1H, dd, J=4.7, 2.2, H-4), 6.16 (1H, d, J=4.7, H-3), 6.21 (1H, d, J=2.2, H-5), 7.23 (2H, d, J=8.9, H_{A-2,6}), 7.35 (2H, d, 8.9, H_{A-3,5}), 7.65 (1H, d, J=8.7, H_{B-2,6}), 7.77 (2H, d, J=8.7, H_{B-3,5}), 8.21 (2H, d, J=8.5, H_{C-2,6}), 8.24 (2H, d, J=8.4, H_{B-3,5}); ¹³C-NMR (d₆-DMSO, δ, 75.5 MHz): 71.5 (C-3), 81.6 (C-5), 98.6 (C-4), 117.4 (C_{A-2,6}), 123.9 (C_{C-3,5}), 124.0 (C_{B-3,5}), 127.3 (C_{A-4}), 128.5 (C_{C-2,6}), 128.7 (C_{B-2,6}), 129.1 (C_{A-3,5}), 142.6 (C_{C-1}), 144.8 (C_{B-1}), 147.0 (C_{A-1}), 147.5 (C_{C-4}), 147.9 (C_{B-4}). *Anal.* Calcd for C₂₁H₁₅N₄O₄Cl: Calc. for C, 53.57; H, 3.21; N, 11.9%. Found: C, 53.72; H, 3.53; N, 9.89%.

REFERENCES AND NOTES

- [1] Banerji, A.; Biswas, P. K.; Bandyopadhyay, D.; Gupta, M.; Prangé, T.; Neuman, A.; *J. Heterocyclic Chemistry*, (Part XI) **2007**, *44*, 137.
- [2] Black; D. St. C. Crozier, R. F.; Davis, V. C.; *Synthesis* **1975**, 205.
- [3] Tufariello, J. J.; *1,3-Dipolar Cycloaddition Chemistry*; John Wiley and Sons, Inc.: New York, 1984.
- [4] Houk, K.N.; Yamaguchi, L. in "1,3-Dipolar Cycloaddition Chemistry", ed. A. Padwa, Wiley-Interscience, New York, Vol. III, p. 407, 1984.
- [5] Banerji, A.; Sahu, A.; *J. Sci. Ind. Res.*, **1986**, *45*, 355.
- [6] B.M. Trost, Ian Fleming, "Comprehensive Organic Synthesis" Pergamon Press, Vols. IV and V, 1991.
- [7] Torsell, K.B.G.; "Nitrile Oxides, Nitrones and Nitronates in Organic Synthesis", VCH, New York, 1988.
- [8] Iralapati, N.R.; Baldwin, J.E.; Adlington, R.M.; Pritchard G.J.; Cowley, A.R.; *Tetrahedron* **2005**, *61*, 1773.
- [9] O'Neil, I. A.; Ramos, V.E.; Ellis, G.L.; Cleator, E.; Chorlton, A.P.; Kalindjian, S. B; *Tetrahedron Lett.*, **2004**, *45*, 3659.
- [10] Banerji, A.; Bandyopadhyay, D.; Prangé, T.; Neuman, A D. Bandyopadhyay; *Tetrahedron Letters*, **2005**, *46*, 2619..
- [11] Saito, A. T.; Yamada, T.; Miyazaki, S.; Otani, T.; *Tetrahedron Lett.* **2004**, *45*, 9581.
- [12] Saito, A. T.; Yamada, T.; Miyazaki, S.; Otani, T.; *Tetrahedron Lett.* **2004**, *45*, 9585.
- [13] Sustmann, R. *Tetrahedron Lett.*, **1978**, *29*, 2717.
- [14] Crystallographic Data for compound **3aa** and **3cb** have been deposited with the Cambridge Crystallographic Data Centre as CCDC 235162 and 248040 respectively. Copies of the data can be obtained, free of charge, on application to CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44 1223 336033 or e-mail:deposit@ccdc.cam.ac.uk).
- [15] G. H. Grant, W. G. Richard, Computational Chemistry. Oxford Chemistry Primers, Oxford University Press, 1995.
- [16] Sheldrick, G. M.; SHELXS-86. Program for Solution of Crystal Structures. University of Göttingen, Germany, 1986.
- [17] Sheldrick, G. M.; Schneider, T. R.; *Methods Enzymol.* **1997**, *277*, 319.